

The structure of an atom

1. Noun
2. Number
3. Noun
4. Noun
5. Verb
6. Plural Noun
7. Plural Noun
8. Noun
9. Adjective
10. Adjective
11. Noun
12. Noun
13. Verb
14. Adjective
15. Noun
16. Noun
17. Noun
18. Noun
19. Number

The structure of an atom

An _____ is so small that _____ of them could fit on the head of a pin. Yet, _____ are made up of even smaller particles called _____ protons, and electrons, as _____ in figure 1. _____ and _____ are located at the center of the atom, which is called the _____.

Protons are _____ charged particles and neutrons are particles that have no charge. Electrons are _____ charged particles that are located outside the _____. _____ constantly move _____ an atom's nucleus in energy levels. The basic structure of an atom is the result of the _____ between _____ and _____. _____ contain an equal number of protons and electrons, so the overall charge of an _____ is _____.