

The structure of an atom.

1. Noun
2. Verb
3. Preposition
4. Adjective
5. Verb
6. Noun
7. Article
8. Verb
9. Noun
10. Noun
11. Verb
12. Noun
13. Noun
14. Noun

The structure of an atom.

An _____^{Noun} is so small that billions of them could _____^{Verb} on the head of a pin. _____^{Preposition}
atoms are made up of even _____^{Adjective} particles called neutrons, protons, and electrons. Neutrons and
protons are _____^{Verb} at the center of the _____^{Noun} which is called _____^{Article} nucleus. Protons are
positively _____^{Verb} particles, and neutrons are particles that have no charge. Electrons are negatively
charged particles that are located outside the _____^{Noun}. _____^{Noun} constantly _____^{Verb} around an
atom's nucleus in energy levels. The basic structure of an _____^{Noun} is the result of the attraction between
_____^{Noun} and _____^{Noun}. Atoms contain an equal number of protons and electrons, so the overall
charge of an atom is zero.